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Stoichiometry

Stoichiometry Name _____ Per _____
 $2\text{MnO}_2 + 4\text{KOH} + \text{O}_2 + \text{Cl}_2 \rightarrow 2\text{KMnO}_4 + 2\text{KCl} + 2\text{H}_2\text{O}$
9. For the reaction above, there is 100. g of each reactant available. Which reagent is the limiting reagent? [Molar Masses: MnO

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26. According to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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Course helps students learn the essential chemistry lessons of stoichiometry.

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